Concise and Efficient Synthesis of Dehydropeptide Analogues from Unsaturated 5(4H)-Oxazolone

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Abstract: Some dehydropeptide analogues were directly synthesized by the reaction of unsaturated oxazolones with free amino acids.

Keywords: Azlactone, free amino acid, dehydrodipeptide.

For improving the bioactivity and bioavailability of peptides, one of the approaches is to introduce a dehydroamino acid residue into the peptide sequence without adversely affecting the bioactivity¹. The Erlenmeyer reaction² was most frequently employed to prepare dehydroamino acids with aromatic and heterocyclic substituent³.

However, the presence of double bond in a dehydroamino acid dramatically decreases both the nucleophilicity of the amino group and the reactivity of carboxylic group, making the acylation of dehydroamino acids is not a satisfactory method for the preparation of dehydropeptide⁴.



The unsaturated oxazolone can be converted into the corresponding dipeptide by two reaction sequences (Scheme 1). One is to condense 1 with the amino acid ester to form 2. The resulting dipeptide ester 2 was hydrolyzed to afford 3^5 . Another reaction sequence is direct condensation of 1 with free amino acid without esterification and saponification. This method would be more favorable.

In the present study, the possibility of ring-cleavage of oxazolone **1** with free amino acid was explored. Considering the reluctant solubility of free amino acid in organic solvents, the water mixing organic solvents like acetone, pyridine, dioxane,

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tetrahydrofuran were used as reaction media. Among them, the mixed solvent of THF-water (4:1) has been proved particularly efficient to dissolve all related reactants and to give reasonable yields of products ($68 \sim 80\%$).

Table 1 The physical and chemical data of the target compounds 3a~f



Experimental

To a solution of oxazolone 1^4 (1 mmol) in THF (5 ml) was added a suspension of amino acid (1.1 mmol) and TEA (1.1 mmol) in THF/H₂O (7:3) (10 ml). After 12 hours of stirring at room temperature, the solvent was removed in vacuum and the residue was partitioned between 10% Na₂CO₃ and diethyl ether. The aqueous layer was washed twice with ether, then chilled to 0° C and acidified to pH 2 with saturated KHSO₄ solution. The precipitate was filtered, washed with cold water, and recrystallized from ethonal/ H₂O (1:1).

References

- 1.
- M. L. English, C. H. Stammer, Biochem. Biophys. Res. Commun., 1978, 85, 780. Vogel's Textbook of Practical Organic Chemistry, 4th edition, Longman Inc. New York 1978, 2. n.885.
- U. Schmidt, E. Ohler, J. Hausler, et al., Progress in the chemistry of organic natural products, 3. Vol. 37, Springer-Verlag, Wien, **1979**, p.251. C. Shin, Y.Yonezawa, T.Yamada, *Chem. Pharm. Bull.*, **1984**, *32*, 2825.
- 4.
- 5. J. V. Edwards, A. R. Lax, Int. J. Peptide & Protein Res., 1986, 28, 603.

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